

Key

Additional Problems: Oxygen Transport

Please use in room 153. You may borrow for up to 1 hour for copying.

Thales
AR

$$1) Y = \frac{pO_2^n}{pO_2^n + P_{50}^n} \quad n=1, P_{50}=1^{\circ} \text{ torr}, pO_2 = 2.43 \text{ torr}$$

$$Y = \frac{2.43}{2.43 + 1.00} = \underline{0.708}$$

$$\text{or } pO_2 = 5.5 \text{ torr}$$

$$Y = \frac{5.5}{5.5 + 1.0} = \underline{0.85}$$

$$2) n=2.8, P_{50} = 26 \text{ torr } pO_2 = 15.0 \text{ torr or } 87.5 \text{ torr}$$

$$Y = \frac{(15.0)^{2.8}}{(15.0)^{2.8} + (26)^{2.8}} \quad Y = \frac{(87.5)^{2.8}}{(87.5)^{2.8} + (26)^{2.8}}$$

$$Y = \underline{0.18} \quad Y = \underline{0.97}$$

$$3) \Delta Y = Y_{\text{lungs}} - Y_{\text{tissues}} = \frac{100^{2.8}}{100^{2.8} + 26^{2.8}} - \frac{20^{2.8}}{20^{2.8} + 26^{2.8}}$$

$$= 0.98 - 0.32 = \underline{0.66}$$

b) 100 molecules of Hb is 400 subunits or 400 O_2 -binding sites. On average, 98% of these sites will fill in the lung, + only 32% will remain filled after passing through tissues. So the capacity to deliver O_2 is 66% of the number of carrier sites: $400 \times .66 = 264 O_2$ molecules. 264 mmole O_2 delivered per 100 mmole Hb tetramer.

$$4. a) \text{noncooperativity} \Rightarrow n=1 \quad \Delta Y = \frac{100}{100+26} - \frac{20}{20+26} = \underline{0.35}^{\#}$$

b) Mutations at subunit interfaces might affect cooperativity. But because of the complexity of the cooperative mechanism, it's hard to rule out other sites. Probably not surface residues or heme-binding residues

* transport reduced by about half

5. a) $n = 2.8$ $P_{50} = 10$ torr

$$\Delta Y = \frac{100^{2.8}}{100^{2.8} + 10^{2.8}} - \frac{20^{2.8}}{20^{2.8} + 10^{2.8}} = 0.99 - 0.87 = \underline{0.12}$$

(transport reduced by about 80%)

b) Mutation at BPG binding site might lower affinity for BPG, thus lowering P_{50} .

6. See next page

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6. a) $Y_A = \frac{a}{a+P_{50}}$ $Y_B = \frac{b}{b+P_{50}}$ Let $P = P_{50}$ (to save writing)

$$\Delta Y = \frac{a}{a+P} - \frac{b}{b+P} = \frac{a(b+P) - b(a+P)}{(a+P)(b+P)}$$

now, simplify:

$$\Delta Y = \frac{ab + aP - ab - bP}{ab + aP + bP + P^2} = \frac{P(a-b)}{P^2 + P(a+b) + ab}$$

take derivative with respect to P . (a and b are constants)

$$\frac{d\Delta Y}{dP} = \frac{(P^2 + P(a+b) + ab)(a-b) - P(a-b)(2P + a+b)}{(P^2 + P(a+b) + ab)^2} \quad *$$

If $\frac{x}{y} = 0$ then $x = 0$, so $\frac{d\Delta Y}{dP} = 0$ if the numerator = 0.

$$(a-b) [P^2 + P(a+b) + ab - 2P^2 - aP - bP] = 0$$

if $a \neq b$, then $a-b \neq 0$, so term in brackets = 0.

(If $a=b$, no net transport anyway.)

$$P^2 + P(a+b) + ab - 2P^2 - P(a+b) = 0$$

$$\text{or } -P^2 + ab = 0$$

$$\text{or } P^2 = ab \quad P = \sqrt{ab}.$$

b) For lump-to-lumps, $a = 100$, $b = 20$.

$$P = \sqrt{100 \cdot 20} = 45$$

The P_{50} of Hb may be a compromise between what is optimal for 100-ton-to-20-ton transport + other functions, such as response to pH change.

$$* \text{ Because } \frac{d}{dx} \left(\frac{f(x)}{g(x)} \right) = \frac{g(x) \frac{d}{dx} (f(x)) + f(x) \frac{d}{dx} (g(x))}{(g(x))^2}$$